

Thin-layer hydration method to prepare a green tea extract niosomal gel and its antioxidant performance

Original Paper

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Abstract This study aimed to prepare a niosomal gel of green tea (*Camellia sinensis*) extract containing catechins, mostly epigallocatechin-3-gallate (EGCG), as a potent antioxidant. Niosomes can increase EGCG's stability and penetration into the skin for a better therapeutic effect. Niosomes were prepared by a thin-layer hydration method, were evaluated for their vesicle shape, particle size, polydispersity index, zeta potential and entrapment efficiency, and then incorporated into gels using sodium alginate as a gelling agent. Three niosomal gel formulations were prepared with different concentrations of niosomes green tea extract. Afterwards, organoleptic properties, chemical and physical characteristics, antioxidant activity, and stability and irritability of the niosomal gels were investigated. The different concentrations of green tea extract had a significant effect on the physical characteristics, but not on the chemical ones. Its antioxidant activity was determined using the 1,1-diphenyl-2-picrylhydrazyl (DPPH) scavenging method. The 50% extract green tea niosomes gel showed the highest inhibition value (25.13%). The stability was determined by freeze-thaw and real-time methods; they showed a decrease in pH, but still within the pH range of skin. The irritability test used was the Hen's Egg Test-Chorioallantoic Membrane (HET-CAM) method, which showed no irritation for all formulas. In conclusion, 50% green tea extract niosomes gel results showed it to be the best formulation with optimal antioxidant results.

Keywords green tea extract – niosomes gel – antioxidant activity – stability – irritability

INTRODUCTION

Ageing is a multifactorial process that causes structural and physiological changes in each layer of the skin, such as fine lines and wrinkles, depigmentation, pale skin colour, change in texture and the loss of elasticity (Draelos, 2016). Ageing is influenced by endogenous and exogenous factors (Ramos-e-Silva et al., 2013). Extrinsic skin damage can develop due to several factors, and it is thought that among the environmental factors, UV rays are usually the most damaging (Perricone, 2008). UV induces the generation of reactive oxygen species (ROS) in the skin; oxidative stress conditions develop when the ROS formation exceeds the antioxidant defence ability of the affected cells (Poljšak & Dahmane, 2012). One strategy to reduce the risk of UV rays is the use of exogenous antioxidants, which could be administered orally or topically (Pandel et al., 2013).

Green tea (*Camellia sinensis*) contains bioactive chemicals and it is particularly rich in the flavanol group of polyphenols known as catechins, which may constitute up to 30% of dry leaf weight (Graham, 1992). Epigallocatechin-3-gallate

(EGCG) and epicatechin-3-gallate (ECG) are the most effective antioxidant compounds found in green tea (Namal Senanayake, 2013). EGCG is easily oxidated and unstable when heated or exposed to light (Widyaningrum et al., 2015). EGCG has a low skin-penetrating ability because of its chemical interaction with the skin lipid bilayer (Wisuitiprot et al., 2011). It is soluble in water or has high hydrophilicity, making it difficult to penetrate the cell membrane (Song et al., 2014). Therefore, it is necessary to create a delivery system that can protect EGCG from degradation as well, so that it will increase the green tea extract's effectiveness as an antioxidant.

Niosomes are vesicle systems that are amphiphilic in nature and entrap hydrophilic drugs in the core cavity and hydrophobic drugs in the non-polar region present within the bilayer; hence, both hydrophilic and hydrophobic drugs can be incorporated into the niosomes (Shakya & Bansal, 2018). Niosomes exhibit similar physical-chemical properties to liposomes, but the bilayers are formed by nonionic

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surfactants. Compared with liposomes, niosomes are highly stable for a longer period (Kumar, 2019; Bartelds et al., 2018). The vesicles can work as a depot to release the drug slowly, providing a controlled release (Sharma et al., 2018). Niosomes also increase drug stability, accelerate the therapeutic effect and increase the drug's penetration ability when applied topically (Muzzalupo & Tavano, 2015). The bilayer structure of niosomes can protect the enclosed active pharmaceutical ingredient from heterogeneous factors, so it can be used for labile and sensitive drugs. Niosomes are suitable for encapsulating a variety of active pharmaceutical ingredients (Sankhyan & Pawar, 2012).

Topical gels have the advantage of being easily applied, distributing evenly to the skin, easily drying, giving a cold feeling to the skin and forming a layer that is easily washed away as needed (Rathod & Mehta, 2015). Furthermore, the physicochemical properties of niosomes are maintained (Rajkumar et al., 2019). Also, because of their high water content and swelling processes, they can offer a better feeling for the skin. Furthermore, gel preparations are suitable for drugs with hydrophilicity (Silna et al., 2016).

In this study, a topical antioxidant preparation was made with the niosomes system and incorporated into the gel with sodium alginate as the gelling agent to overcome the absorption and stability problems of EGCG in green tea extract. Alginate is widely used as a hydrogel agent in biomedicine, including drug delivery applications. It is widely used due to its biocompatibility, low toxicity, relatively low cost and mild gelation (Lee & Mooney, 2012).

MATERIALS AND METHODS

Materials

Niosome preparation

The materials used were cholesterol (Sigma-Aldrich); sorbitan monostearate (Span 60) from CRODA Health Care; polysorbate 60 (Tween 60) from Matching Nature; calcium chloride (CaCl_2) and polyethylene glycol 400 (PEG 400) from Merck; methylparaben, sodium benzoate, propylene glycol, carboxymethyl cellulose sodium (CMC-Na) from Bratacochem; sodium alginate from Jiejing Shadong Corporation; methanol (ProAnalytics); 0.9% sodium chloride (NaCl), distilled water, sodium lauryl sulphate (SLS) and hen's egg (for irritation testing) purchased from an authentic source; 2,2-diphenyl-1-picrylhydrazyl (DPPH) (Sigma-Aldrich); green tea extract (catechin 98%) from In Health.

Green tea extract niosome characterisation

The green tea extract niosomes were characterised based on the vesicle shape, particle size, distribution (polydispersity index [Pdl]), zeta potential and entrapment efficiency. Green tea extract niosomes samples were prepared by spreading

a niosomes suspension with 5% CMC-Na on an object glass until a dry layer was formed. The composition of green tea (*C. sinensis*) extract niosomes is reported in Table 1.

Vesicle shape

The morphological characterisation of niosomes was determined using Scanning Electron Microscopy (SEM) at the magnifications of 1000 \times , 2000 \times and 5000 \times .

Niosomes particle size, Pdl and zeta potential

The particle size, Pdl and zeta potential were determined using Zetasizer Nano (Malvern Ltd.).

Entrapment efficiency (EE)

Entrapment efficiency was determined by calculating the amount of entrapped drug. As much as 100 mg niosomes of green tea extract was hydrated with 10 mL of phosphate buffer saline solution and sonicated in a bath sonicator for 10 minutes. The green tea extract containing niosomal dispersion was separated by centrifugation at 2500 rpm for 45 minutes. The clear supernatant was analysed using a UV spectrophotometer at λ_{max} of 273 nm to calculate the amount of entrapped drug. All assessments were completed in triplicate. The percentage of drug encapsulation was calculated by the following equation (Patil et al., 2017; Khan & Irchhaiya, 2020):

$$EE(\%) = \frac{(C_t - C_f)}{C_t} \times 100 \quad (1)$$

where C_t is the concentration of the total drug and C_f is the concentration of the free drug.

Preparation of green tea extract niosomes gel

To prepare the green tea extract niosomes gel, the proportions of sodium alginate (as a gelling agent), CaCl_2 (water absorbing agent to improve the gel properties), propylene glycol (humectant), methylparaben (preservative) and purified water were measured by weight and they are given in Table 2. The gel was prepared by dissolving methylparaben in propylene glycol (mixture A). Sodium alginate was crushed until smooth and put into water. After being hydrated, it was homogenised while adding CaCl_2 (mixture B). Mixture A was added to mixture B and then stirred with a homogeniser at 1000 rpm at room temperature until homogeneous. The niosomes were suspended until homogeneous, after which the remaining water and fragrance were added to form a consistent gel (Isnan & Jufri, 2017).

Table 1. Composition of green tea (*C. sinensis*) extract niosomes.

Material	Composition
Span 60	12.6 gram
Tween 60	19.62 gram
Cholesterol	1.74 gram
PEG 400	90.0 gram
Green Tea Extract	60.0 gram
Sodium Benzoate	0.6 gram
Distilled water	to 600 ml

Table 2. The composition of green tea extract niosome gel.

Material	Composition (%) (w/w)		
Green tea extract niosome	30	40	50
Sodium alginate	5	5	5
CaCl ₂	0.2	0.2	0.2
Propylene glycol	10	10	10
Methylparaben	0.1	0.1	0.1
Green tea fragrance	sufficiently	sufficiently	sufficiently
Distilled water	to 100	to 100	to 100

Green tea extract niosomes gel evaluation

The evaluations of the green tea extract gel with niosomes system were carried out based on the organoleptic test, physical (viscosity, spreadability) and chemical (pH) characteristics tests, physical stability tests (high temperature, room temperature, low temperature and cycling test), as well as the antioxidant and irritability tests.

Organoleptic test

The organoleptic test was descriptive, done by observing the odour, colour and consistency of the preparation visually (Sharif et al., 2017).

Determination of pH

The pH measurement was determined using a pH meter. The pH meter was calibrated with a pH 7.0 standard buffer solution before measurement. The electrodes of the pH meter were dipped into 10 g of each gel preparation (Aiyalu et al., 2016).

Determination of viscosity

The viscosity was determined using a Brookfield viscometer. A total of 100 g of green tea extract gel with niosomes system was put into a beaker glass; then, a size 64 spindle was

installed on the viscometer and lowered until immersed into the preparation. The rotor ran at 1.5 rpm until the viscometer showed a stable number. Then, this value was recorded and multiplied by the correction factor according to the spindle size (Singh et al., 2013).

Spreadability test

Measurements were done by placing 0.5 g of green tea extract gel with niosomes system in a pair of transparent glass plates. Then, different weight loads (0, 50, 100, 150 and 200 g) were placed on the top of the glass plate for 1 minute per load (Dantas et al., 2016).

Real-time stability test

The ability of the vesicles to retain the desired characteristics of the niosomes was determined by storing the niosome suspension under different conditions: 4 °C ± 2 °C (refrigerator), 30 °C ± 2 °C (room temperature) and 40 °C ± 2 °C (climatic chamber) for 1 month. A total of 10 g of the sample was placed into vials and stored at different temperatures. Organoleptic and pH values of the sample were tested on the first day (1st day) just before storage and the last day of storage (30th day) (Kumar & Dua, 2018).

Accelerated stability test (freeze–thaw cycle)

A total of 10 g of each sample was placed into vials and stored at 4°C ± 2°C for 24 hours and then transferred to 40°C ± 2°C for 24 hours (counted as one cycle). The process was repeated up to six cycles and the organoleptic and pH values were evaluated at the end of the cycle (WHO, 1996).

Antioxidant activity test

Antioxidant activity was measured using the DPPH scavenging method for green tea extract, green tea extract gel and green tea extract gel with the niosomes system. (1) *Green tea extract*: Firstly, green tea extract solutions were prepared at 5, 10, 20, 50 and 100 ppm concentrations. The observations were carried out using a UV–Vis 1240 spectrophotometer at 517-nm wavelength and using vitamin C as an antioxidant positive control. (2) *Green tea extract gel and green tea extract gel with niosomes system*: Three 0.5-g samples of each gel containing 30%, 40% and 50% (green tea extract/green tea extract niosomes) were dissolved in water until 10 mL and stirred using ultrasonic stirring for 10 minutes. Then, the samples were dissolved in methanol until 100 mL concentration was reached and stirred with ultrasonic stirring for 20 minutes. The solutions were incubated for 30 minutes at 37 °C. The absorbance was measured using a UV–Vis spectrophotometer 1240 at 515–517 nm wavelength (Gane & Parki, 1958). % Antioxidant activity was calculated by using the equation as follows:

Table 3. The characteristics of green tea extract niosomes obtained from Malvern Zetasizer Nano.

Niosome	Z-Average diam. (nm)	Polydispersity Index	Zeta potential (mV)	Entrapment efficiency (%)
Mean	1109	0.856	-0.616	89.74
Standard Deviation	± 63	± 0.057	± 0.068	± 2.52

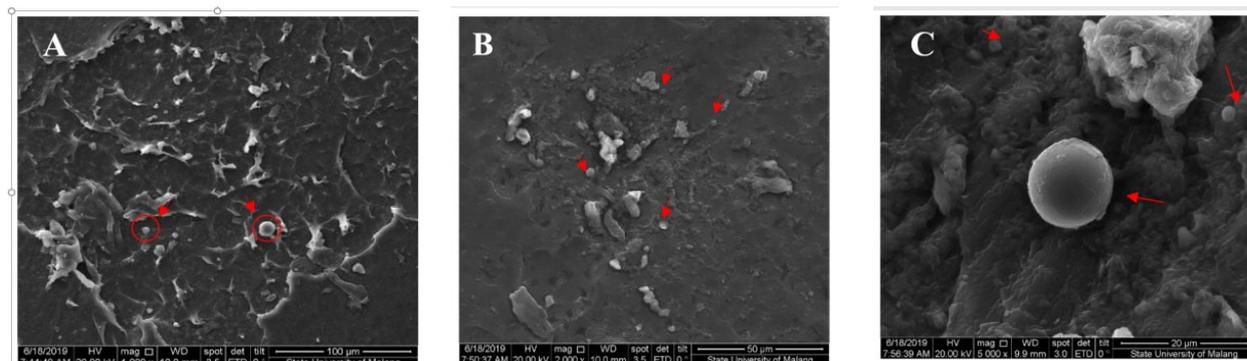


Figure 1. Scanning electron micrographs of the morphology of green tea extract niosome system at a magnification of (A) 1000×; (B) 2000× and (C) 5000×.

$$\% \text{ antioxidant activity} = (A_0 - A_1)/A_0 \times 100 \quad (2)$$

where A_0 is the absorbance of free radical (DPPH solution) and A_1 is the absorbance of the antioxidant (sample).

Irritability test

Irritability was tested using the Hen's Egg Test Chorioallantoic Membrane (HET-CAM) method. White Leghorn eggs were prepared, put in incubator trays at 37 °C and then rotated for 10 days. On the 10th day of incubation, the eggshell was softened by sterile 0.9% NaCl and scratched around with sterile scissors. The outer membrane was moistened with a warm 0.9% NaCl solution and they were put again in the incubator trays for 5–20 minutes; then, the inner egg membranes were removed to expose the vascular Chorioallantoic Membrane (CAM). After application of the test substance, about 300 g sample (niosomes gel of green tea extract) was put into CAM for 20 seconds and then cleaned up immediately using sterile 0.9% NaCl solution. The observations were started after the CAM was cleaned; the samples were examined for about 300 seconds and then scored for irritant effects (lysis, haemorrhages and coagulation). SLS was added as a positive control and distilled water as a negative control (Luepke, 1985).

Statistical analysis

The data of pH, viscosity and spreadability test were analysed using the Shapiro–Wilk test. A p -value >0.05 indicated a normal distribution, allowing the analysis to continue with

a one-way analysis of variance (ANOVA) to find a significant difference between the formulas. The test results continued with the Tukey's Honestly Significant Difference (Tukey HSD) test with a confidence level of 95% ($\alpha = 0.05$). If the pH, viscosity and spreadability test had a p -value <0.05 , the variance test was performed with the Kruskal–Wallis test. These analyses were performed using the Statistical Package for the Social Sciences (SPSS) v16.0 program (Chicago, USA).

RESULTS AND DISCUSSION

Niosomes characterisation

The green tea extract niosomes characteristics were observed, including the niosomes morphology (vesicle shape), particle size, distribution (PdI), zeta potential and entrapment efficiency. The morphology is shown in Fig. 1 and the characteristics are given in Table 3.

SEM observation of niosomes at 5000× magnification showed that the niosomes were round in shape (spherical) with a smooth surface and a visible vesicle. Using the Malvern Zetasizer Nano, it was found that the average size of the majority of the green tea extract niosomes vesicles was 1109 nm or 1.109 μm and the particle distribution showed a PdI of 0.856 (Table 3). These niosomes can be classified as large unilamellar vesicles because the size of the vesicles was more than 0.10 μm (Sudheer & Kaushik, 2015). It was also found that the particle size distribution of green tea extract niosomes was very broad, because they had a PdI value >0.7 (Nidhin et al., 2008).

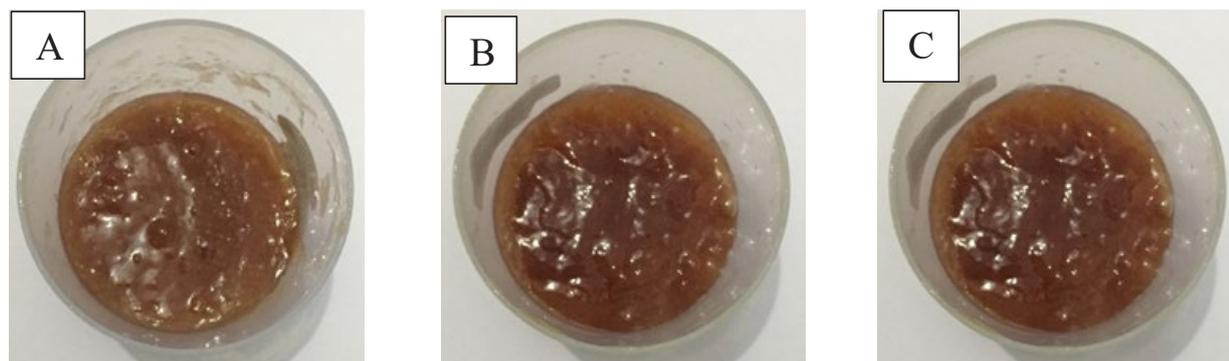


Figure 2. The organoleptic of niosome gel preparations at a niosome concentration of (A) 30%, (B) 40%, and (C) 50%.

The niosomal suspension had a zeta potential of -0.616 mV. This was in the range of -30 to $+30$ mV, so it showed that the niosomes of green tea extract were capable of fast coagulation or flocculation (Sreeram et al., 2008; Okore et al., 2011). The result of entrapment efficiency determination given in Table 3 indicates that the green tea extract niosomes had a good entrapment efficiency with a value of $>80\%$ (Annisa et al., 2016). This might be due to the presence of Span 60, which has the best entrapment efficiency compared with the other grades of Span (Patil et al., 2017; Khan & Irchhaiya, 2020; Asthana et al., 2016; Jaiswal et al., 2016). It can be explained by the structure, orientation and packing behaviour of the surfactant. Span 60 has the longest saturated alkyl chain that increases the entrapment efficiency. Span 60 also has a solid nature, high phase transition temperature and high lipophilicity, which provides a less leaky system and high entrapment capability (Khan & Irchhaiya, 2020; Jaiswal et al., 2016). The presence of cholesterol also makes the niosomes more stable (Patil et al., 2017). Increasing the cholesterol content will increase the hydrophobicity and stability of bilayer vesicles and decrease the permeability (Asthana et al., 2016). The system will be more intact and ordered as a barrier for drug release and also decrease drug leakage by improving the fluidity of the bilayer membrane (Jaiswal et al., 2016).

This study was carried out to experimentally determine the effect of various levels of green tea extract niosomes on drug delivery system characteristics. Niosomes with Span 60, Tween 60, PEG 400 and cholesterol as the main ingredients were made using the thin-film hydration method. The combination of Tween 60 and Span 60 for niosomes formulations can increase the efficiency of drug absorption and release because Tween 60 and Span 60 have chains that interact together (become entangled) and lead to better absorption (Naderinezhad et al., 2017). PEG 400 was added to help to regulate the niosomes system, so that the hydrophilicity and hydrophobicity of the system could be maintained at a balance that positively affects the absorption rate. Low concentrations of PEG 400 cause almost all the PEG 400 molecules to interact with the Span molecule through hydrogen bonds and hydrophilic action to form the

amphiphilic bilayers from proteins (Hua & Liu, 2007). PEG-coated vesicles trap some drugs in long PEG chains, thereby reducing the tendency for particle size to increase (Karin et al., 2010). Also, PEG reduces aggregation, increase stability in vivo and in vitro, and increases the entrapment efficiency and the transformation temperature (T_c) (Hua & Liu, 2007). The addition of cholesterol molecules to the niosomes formula makes the membrane stiffer and reduces the leak of the active ingredients from the niosomes system (Karin et al., 2010).

Organoleptic test

The green tea extract gel with the niosomes system had a thick consistency, was brown in colour and had a distinctive fragrance of green tea. The higher the niosomes level, the stronger was the intensity of the brown colour as shown in Fig. 2.

Physical and chemical characteristics

The results of viscosity, pH value and spreadability of the green tea extract gel preparations with the niosomes system are shown in Table 4.

Different levels of green tea extract niosomes (30%, 40% and 50%) had different viscosities. As can be seen in Table 4, the higher the concentration of green tea extract niosomes, the higher was the viscosity of the prepared gel. The pH was not significantly different and was within the pH range of skin tolerance (4.0–7.0) (Lambers et al., 2006). The spreadability of 30% green tea extract niosomes gel was the highest, which was 0.0051 g/cm; it was lower in the 40% green tea extract niosomes gel (0.050 g/cm), and 50% green tea extract niosomes gel showed the lowest value (0.0040 g/cm). This is consistent with the theory that the greater the viscosity, the lower is the spreadability (Deuschle et al., 2015). Rheological properties of green tea extract niosomes gel 30%, 40% and 50% were evaluated using the data from various speeds of viscosity. The rheogram (Fig. 3) showed a typical concave curve, which indicated that the system has a pseudoplastic flow.

Table 4. Physical and chemical characteristics of green tea extract niosome gel preparation.

Formula	Viscosity (Cps)	pH	Spreadability (g/cm)
Green tea niosome gel 30%	292.667 ± 1.155	6.68 ± 0.03	0.0051 ± 0.0004
Green tea niosome gel 40%	297.333 ± 1.115	6.59 ± 0.10	0.0050 ± 0.0003
Green tea niosome gel 50%	306.667 ± 2.309	6.68 ± 0.01	0.0040 ± 0.0004

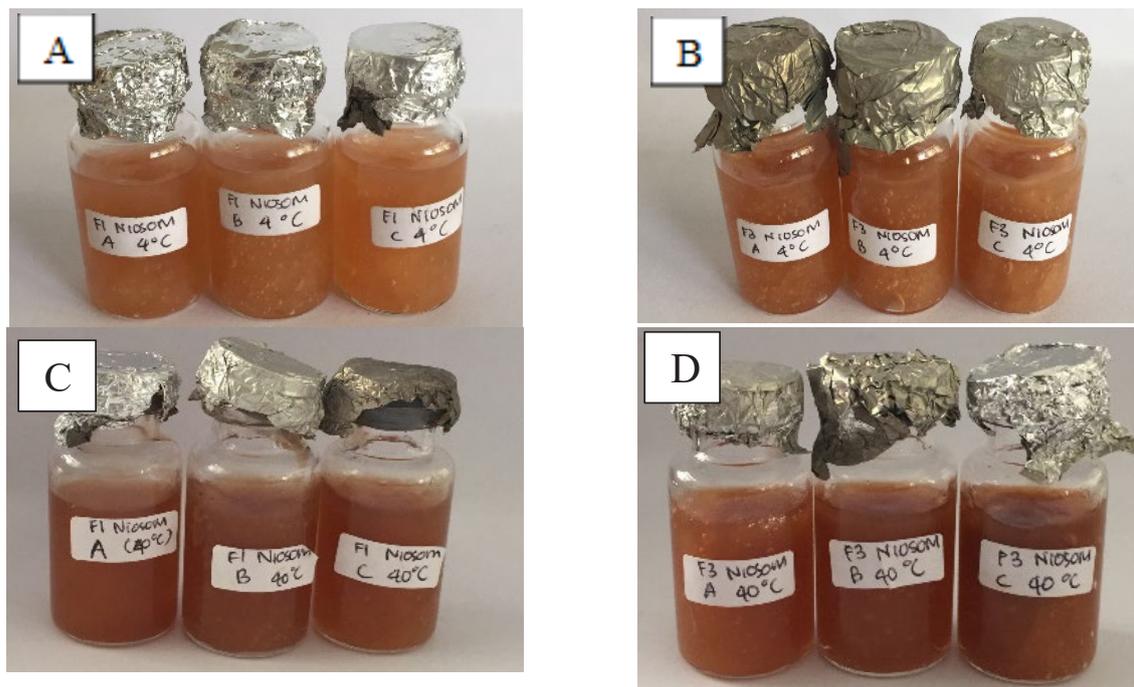


Figure 3. The organoleptic of niosome gel preparation at 4°C on (A) day 1 and (B) day 30; at 40°C on (C) day 1 and (D) day 30 shown the same consistency, brown color and distinctive fragrance of green tea as before storage.

The requirements for a gel base are that it must be inert, safe and not interact with the other ingredients of the formula. The type and concentration of different additives and extracts in the formula will affect the physical and chemical characteristics, stability, effectivity and safety level of the preparation. The base gel to carry the green tea extract niosomes system was sodium alginate. Sodium alginate was chosen because when it is combined with CaCl_2 , it will form a gel without any heating, is solid during the heating process, has a good physical appearance and also has abundant functional groups for a high absorption efficiency (Brovchenko et al., 2005). The function of the PEG 400 additive in green tea extract niosomes sodium alginate during the gel phase was to strengthen the gel structure bonding to form a stronger gel matrix due to increase in viscosity level. The results of TEM (Transmission Electron Microscope) observations showed spherical niosomes morphology with a smooth surface and a diameter of 1109 ± 63 nm (Table 3). Particle morphology is important because less-spherical particles will facilitate contact between the particles and

could lead to aggregation. Niosomes in this study were classified as large vesicles because they were larger than 0.10 μm (Sudheer & Kaushik, 2015). It is a bigger vesicle when compared to the previous study (Isnan & Jufri, 2017). The viscosity test results presented in Table 4 show that the viscosity value of this preparation did not comply with Indonesia National Standards (SNI). According to SNI-16-4399-1996, a good viscosity for the gel is 2.000–50.000 cps. However, these results are similar to those of a previous study; the viscosity was very large due to the addition of gelling agent (sodium alginate) and water absorbing agent (CaCl_2). The pH values of green tea extract niosomes gel formula of different concentrations (30%, 40% and 50%) right after manufacture were acceptable because they were within the pH range of topical preparations (pH 4–8) as stipulated in SNI-16-4399-1996. So, it has a safe level of acidity for use on the skin. These pH values are the same as reported in a previous study (Isnan & Jufri, 2017). To determine the uniformity level of particle size distribution, the PDI was measured. The good particle size distribution was

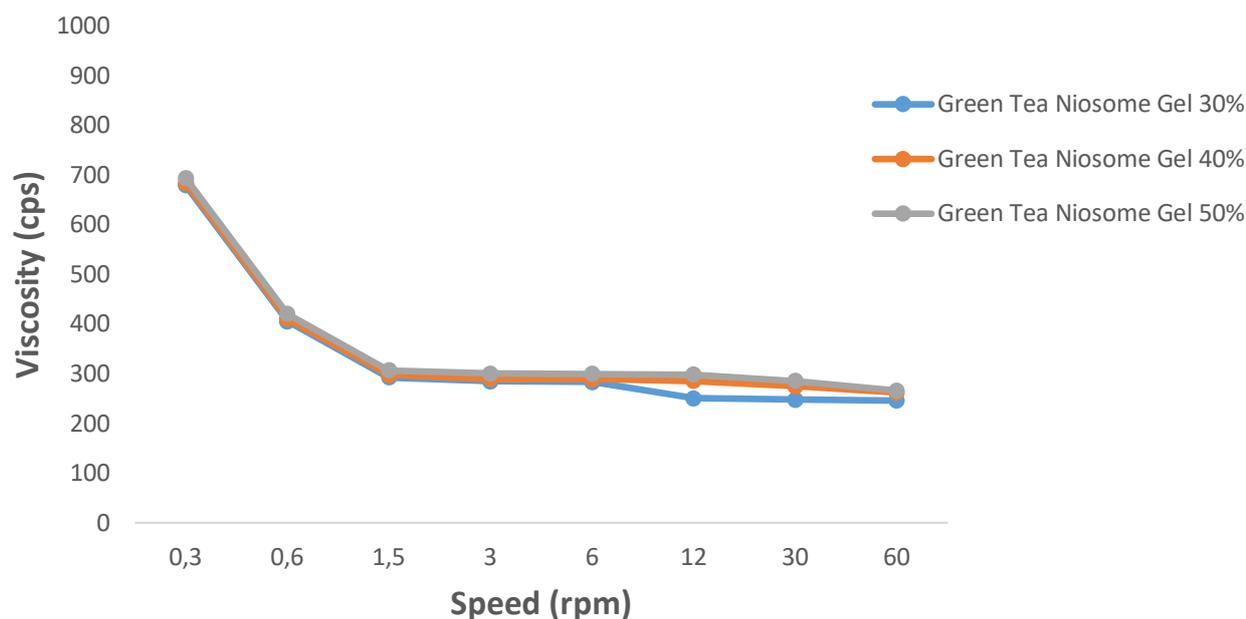


Figure 4. Rheological behavior of green tea extract niosome gel preparation.

Table 5. The pH value of green tea extract niosome gel preparations.

Fomula	1 st day	Freeze-thaw	4°C	30°C	40°C
Green tea extract niosome gel 30%	6.68 ± 0.03	6.16 ± 0.06	6.52 ± 0.02	6.02 ± 0.19	6.01 ± 0.03
Green tea extract niosome gel 40%	6.59 ± 0.10	6.17 ± 0,02	6.52 ± 0.03	6.06 ± 0.04	6.02 ± 0.02
Green tea extract niosome gel 50%	6.68 ± 0.01	6.18 ± 0,10	6.50 ± 0.07	6.04 ± 0.10	5.96 ± 0.06

≤0.3 which shows a homogenous population for colloidal system (Seleci et al., 2016). Particles can be classified into monodispersion groups (Pdl value <0.7) and large size distributions (Pdl values >0.7) (Okore et al., 2011). The niosomes Pdl in this study showed an uneven uniformity of particle size because the Pdl value was ≥0.3. This result was contrary to that of a previous study which showed a Pdl value of 0.349 (Isnain & Jufri, 2017).

Green tea extract niosomes gel stability test

Real-time stability test

It was found that 30%, 40% and 50% green tea extract niosomes gels were the most stable when stored at 4 °C. However, the overall result of organoleptic observations showed that 30%, 40% and 50% green tea extract niosomes gel did not undergo changes in colour, odour and consistency after storage for 30 days and they showed the same organoleptic characteristics as on day 1 (before storage), as can be observed in Fig. 4. Furthermore, the pH values of the gel preparations stored

at 4 °C, 30 °C and 40 °C were decreased slightly over time, as shown in Table 5.

This finding can be due to the low stability of niosomes after manufacture, as shown by the zeta potential value. A sample is called 'stable' if it has a zeta potential more positive than +30 mV or more negative than -30 mV (Surini et al., 2018). The zeta potential of the green tea extract niosomes sample was -0.616 mV (Table 3), which means that the green tea extract niosomes preparation is less stable than found in a previous study in which it was reported as -39.9 mV (Isnain & Jufri, 2017). The zeta potential of green tea extract niosomes was low or close to zero because the constituent ingredients were Tween 60 and Span 60, which are nonionic surfactants. To ensure the preparations retain the same properties after manufacture, the physicochemical characteristics and stability tests were carried out on the green tea extract niosomes gel preparations. The difference between the three formulas was in the colour intensity because green tea extract has a brown colour; the higher the levels of niosomes in the gel base, the browner the colour. Gel preparations had a distinctive smell of green tea because the fragrance of green

tea was used in them. The gel preparation also had a thick consistency as 5% sodium alginate was used as the gelling agent in them. These results are similar to those of a previous study, but the colour was darker due to the increased use of green tea extract (Isnan & Jufri, 2017).

Accelerated stability test

Accelerated stability test using a freeze–thaw cycling test method showed that all the formulas did not change organoleptically. All gels were brown, remained with their typical green tea flavour and had a thick consistency. The pH value measurement results of the freeze–thaw cycling test are shown in Table 5. The three formulas were not significantly different. However, on comparing them to the initial pH values of the preparation before the stability test, the pH values of all gels were found to be decreased.

Niosomal gel stability evaluation showed that the preparation was most stable after 30 days when stored at a low temperature (4 °C), because the reaction slowed at that temperature. But overall, there were no significant changes in pH value, no changes in odour or colour and no visible syneresis, as seen with both the real-time stability and freeze–thaw stability methods (Fig. 4). Overall, this showed that the niosomes system successfully gives a better encapsulation for the active ingredient and increases the stability of preparations.

Antioxidant activity

Before being added to the niosomes system, the green tea extract was examined for its antioxidant activity using the DPPH scavenging method. It was found that the antioxidant activity of green tea extract had an Inhibitory Concentration 50% (IC₅₀) value of 45.56 µg/mL, while the IC₅₀ of vitamin C, used as a positive control, was 5.31 µg/mL. The extract is classified as a very strong antioxidant because it has an IC₅₀ lower than 50 µg/mL (Gane & Parki, 1958). The results of the antioxidant activity of the green tea extract gels and green tea extract niosomes gels are presented in Tables 6 and 7.

The inhibition percentage was obtained from the measurement of 1000 µg/mL samples. The inhibition of 30% green tea extract gel was 17.78% and that of its green tea extract niosomes gel was 10.04%; the 40% green tea extract gel had an inhibition of 19.69% and its green tea extract niosomes gel had a value of 17.49%; the 50% green tea extract gel had an inhibition of 27.09% and its green tea extract niosomes gel had a value of 25.13%. The antioxidant activity increases along with increasing levels of niosomes green tea extract. But it also implies that the addition of green tea extract into niosomes gel has less antioxidant activity compared to conventional green tea extract gel. Besides, the antioxidant activity of green tea extract gel and green tea extract niosomes gel was increased as the concentration of green tea extract niosomes increased (Tables 6 and 7). This finding shows that the niosomes system

Table 6. Antioxidant activity of green tea extract gel.

Formula	% inhibition
Green tea extract gel 30%	17.78 ± 1.49
Green tea extract gel 40%	19.69 ± 1.14
Green tea extract gel 50%	27.09 ± 1.57

Table 7. Antioxidant activity of green tea extract niosome gel.

Formula	% inhibition
Green tea extract niosome gel 30%	10.04 ± 0.05
Green tea extract niosome gel 40%	17.49 ± 0.03
Green tea extract niosome gel 50%	25.13 ± 0.05

provides a high entrapment for active ingredients and its effectiveness as an antioxidant can be decreased due to the ability of the active ingredient to release from the niosomes gel matrix. This study showed a better entrapment compared to the previous study, which reported 77.80% as the highest entrapment efficiency (Isnan & Jufri, 2017).

Irritability test

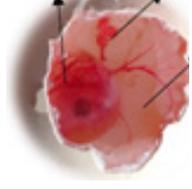
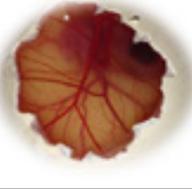
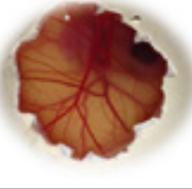
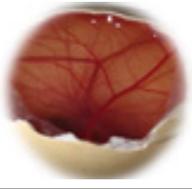
The results of the irritability test are shown in Table 8. SLS as a positive control caused the CAM to haemorrhage in 48 seconds, lysis in 120 seconds and coagulation in 228 seconds. The irritation score was 10.63, which is classified as the strongly irritating category. Distilled water as a negative control caused no haemorrhage, lysis or coagulation on the CAM, which could be classified as the no irritation category. The green tea extract niosomes gel with various levels of green tea extract niosomes showed 0 irritation scores as well, which indicates that the green tea extract niosomes gel did not irritate whatsoever (Luepke, 1985).

Lastly, the irritation test results showed that the green tea extract niosomes gel caused no sign of irritation (haemorrhage, lysis and coagulation) on the chorioallantoic membrane of hen's eggs (Table 8). The niosomes system gave a good entrapment of the active ingredient, which led to better interaction between the green tea extract niosomes gel and the chorioallantoic membrane. No irritation test was conducted in the previous studies of niosomes green tea extract. However, safety testing is essential before raw materials or end products can be sold to customers (Yanti Eff et al., 2019). This test provides practical information on the potential skin damages caused by these topical green tea extract niosomes gel (Suksaeree and Chuchote, 2018).

CONCLUSION

The 30% green tea extract niosomes gel had the best physical properties, showing the highest pH value, viscosity

Table 8. Irritation test results of green tea extract niosome gel

Formula	Treatment		Irritation Score
	Before	After	
SLS (positive control)			10.63 (Strong irritation)
Aquadest (negative control)			0 (No irritation)
Green tea extract niosome gel 30%			0 (No irritation)
Green tea extract niosome gel 40%			0 (No irritation)
Green tea extract niosome gel 50%			0 (No irritation)

and spreadability on human skin. All formulations showed good stability with no change in their organoleptic properties. The 50% green tea extract niosomes gel had the greatest antioxidant activity. Lastly, all formulas also showed no irritation to the hen's egg membrane. The results demonstrated that variation in the concentration of green tea extract niosomes affected the physicochemical characteristics and antioxidant activity, but did not affect the irritability test. We suggest additional evaluation and continuing the stability testing for up to 90 days to improve the preparation of green tea extract niosomes gel. The release of active compounds and penetration evaluation should also be studied to determine the optimum quality of niosomes system.

CONFLICT OF INTEREST

The authors declare no conflict of interest.

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